



Online measurement of dissolved gases in secondary metallurgy

Bartek Wisniewski, Matthias Hölscher, Rajko Antonic,
Torsten Lamp



Online measurement of dissolved gasses in secondary metallurgy

Agenda

1	Introduction
2	Oxygen - Basic principle
3	Oxygen - Application
4	Hydrogen - Basic principle
5	Hydrogen - Application
6	Conclusions

Why to measure dissolved gases in the steel melt?

Oxygen and Hydrogen will degrade material quality

- Higher Oxygen concentration disturbs the solidification process by forming gas cavities (blowholes)
- Oxygen degrades material cleanliness by forming oxides inclusions
- Deoxidation allows to add alloying elements, other case Oxygen will react with additions
- Hydrogen promotes so called Hydrogen Embrittlement of the solidified material, which is very dangerous for special steel grades
- The gas measurement is important to control countermeasures like the level of vacuum degassing



Oxygen Measurement – Basic principle EMF

Electrochemical measurement of electro motive force (EMF)

- In 1889 Walther Nernst described the electrical potential (EMF) as function of a concentration difference:

$$EMF = \frac{R * T}{n * F} * \ln \frac{c_{melt}}{c_{reference}}$$

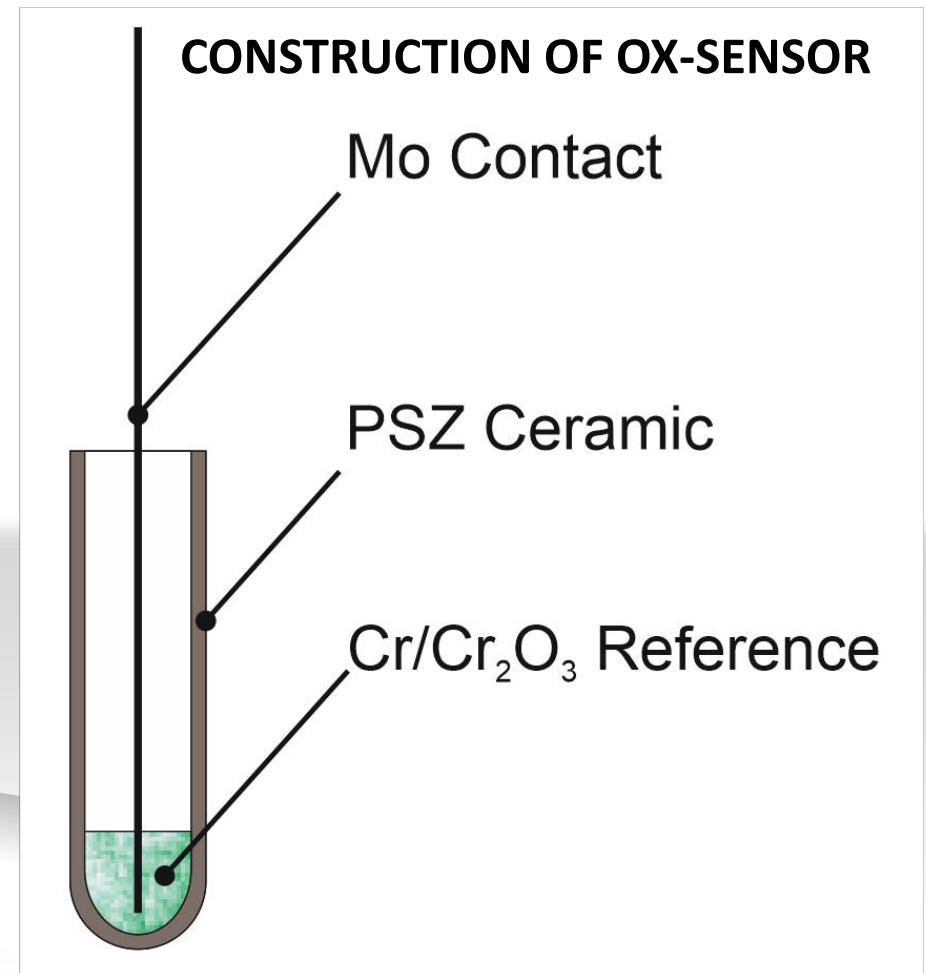
c = Oxygen concentration (= activity)

T = absolute temperature

n = number of charge carrier here $n = 4$

R = universal gas constant $8.314 \text{ J K}^{-1} \text{ mol}^{-1}$

F = Faraday constant $9.6485 \times 10^4 \text{ C mol}^{-1}$



Oxygen Measurement – Basic principle

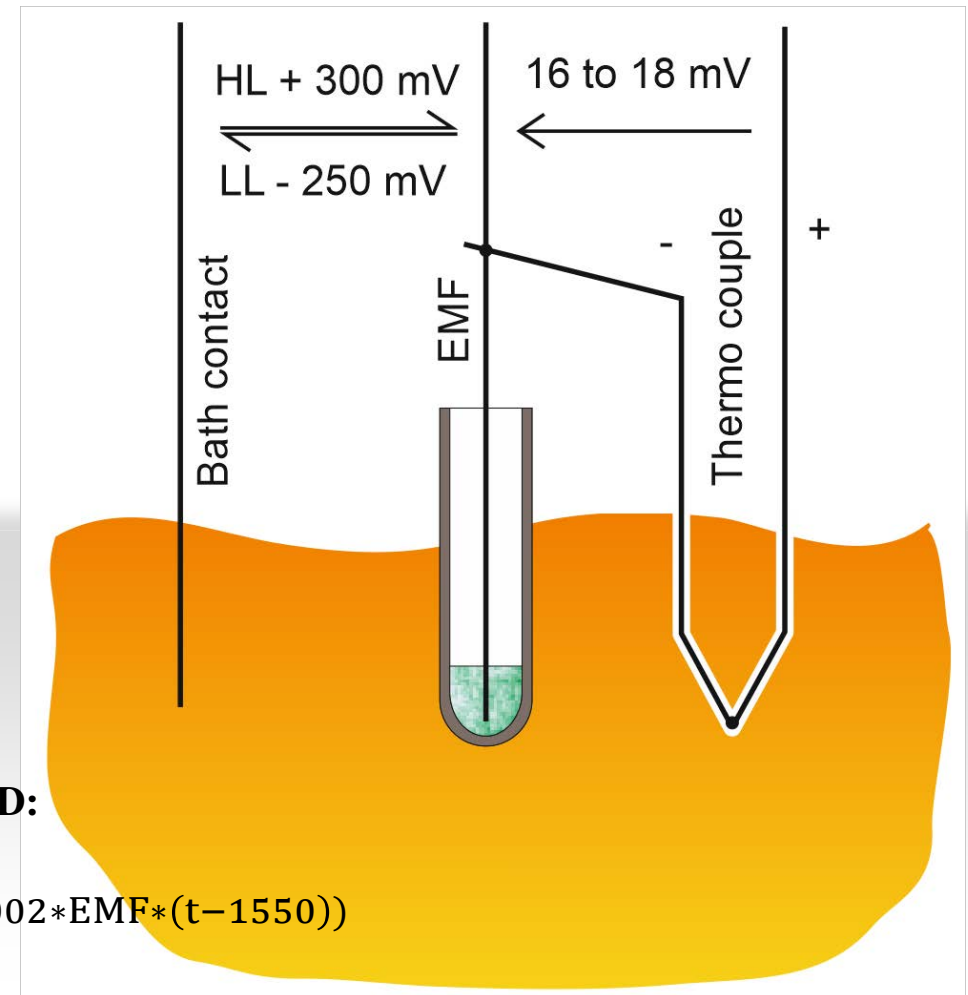
Electrochemical measurement of electro motive force (EMF)

- Temperature measurement is made the same time as EMF measurement. The negative poles are combined.
- EMF represents Oxygen activity difference between reference powder and melt
- In steel two Oxygen levels are present:
 - high level (EMF up to + 300 mV)
 - low level (EMF down to – 250 mV)

OXYGEN CONCENTRATION IN PPM IS CALCULATED:

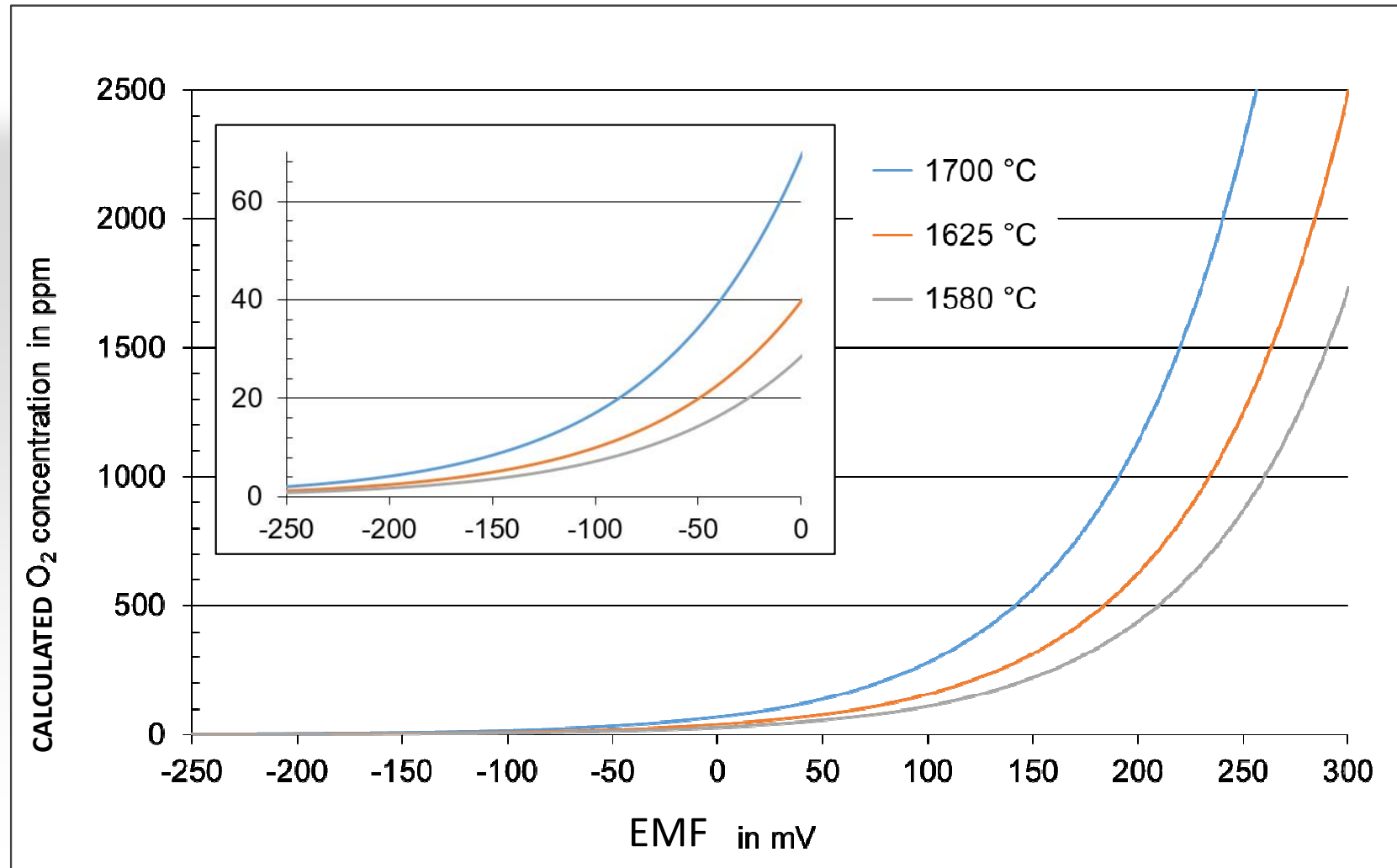
$$C_{\text{Oxygen}} = 10^{1,36 + 0,0059 \cdot (\text{EMF} + 0,54 \cdot (t - 1550) + 0,0002 \cdot \text{EMF} \cdot (t - 1550))}$$

Where: EMF in mV; T in °C



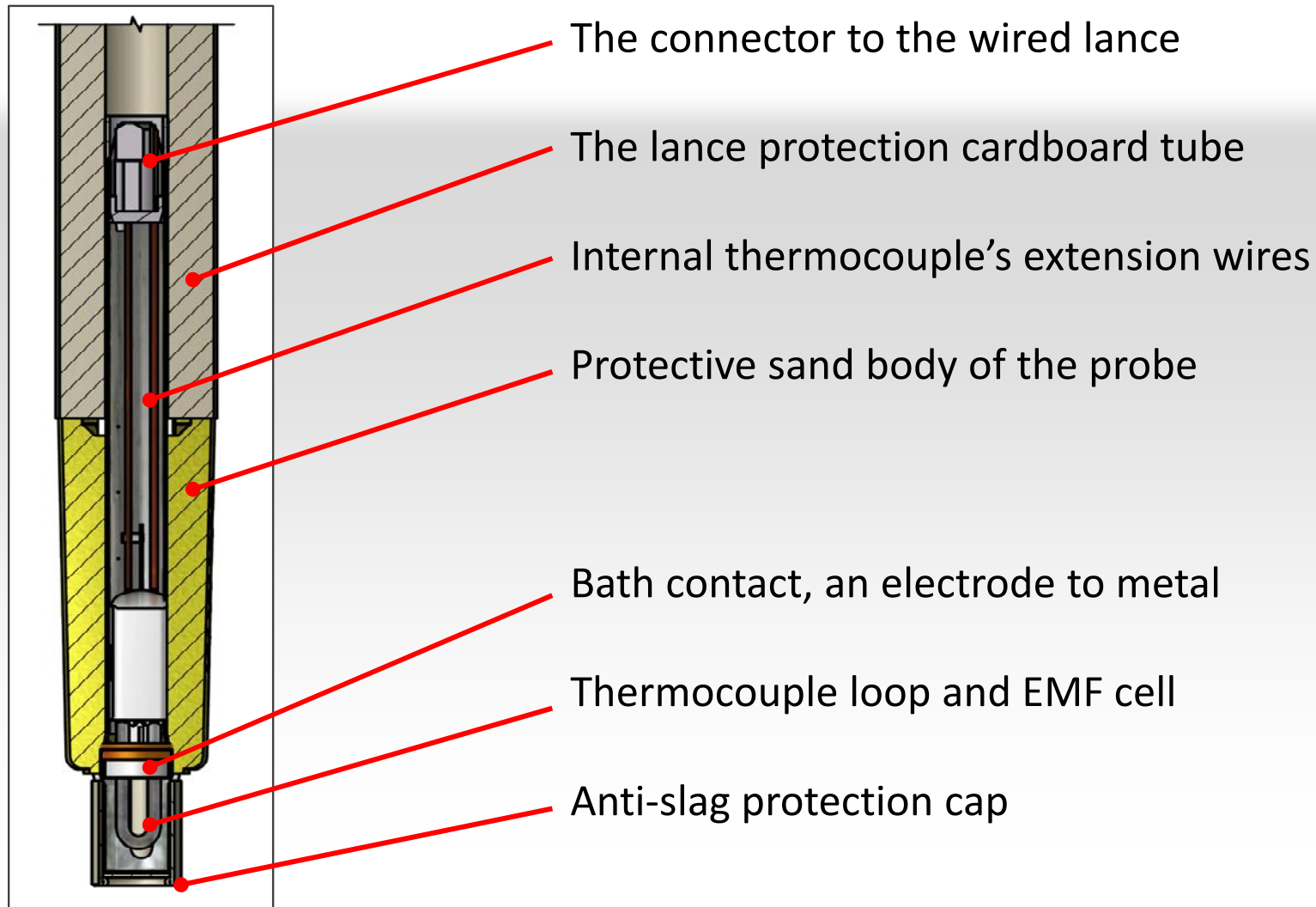
Oxygen Measurement – Basic principle

Exponential influence of melt temperature on concentration determination



Oxygen measurement - application

The construction of the immersion probe



Oxygen measurement - application

Easy to use dipping probes

- Different EMF cell types:
 - high level application, BOF and EAF typically 600 to 1200 ppm
 - low level application, VAC, LF, L, CC typically 1 to 10 ppm
- Used thermocouple types:
 - Type S 1750°C (standard)
 - Type R 1750°C (high accuracy)
 - Type B 1820°C (extended temperature range)
- Cardboard tube protection from 500 to 1750 mm



Oxygen measurement - application

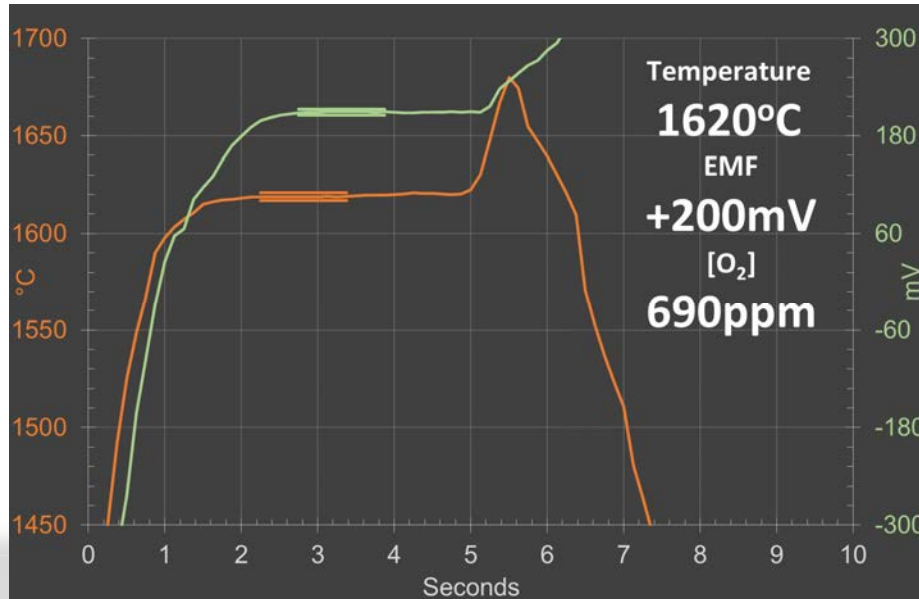
Instrumentation

- Computerized instruments with temperature, EMF recording and implemented functions for %Carbon and %Aluminium for Online calculations
- Typically connected to the steel plant process control system – LEVEL 2
- The very common equipment in every steel shop
- Additional equipment of:
 - Calibration units
 - Spare parts



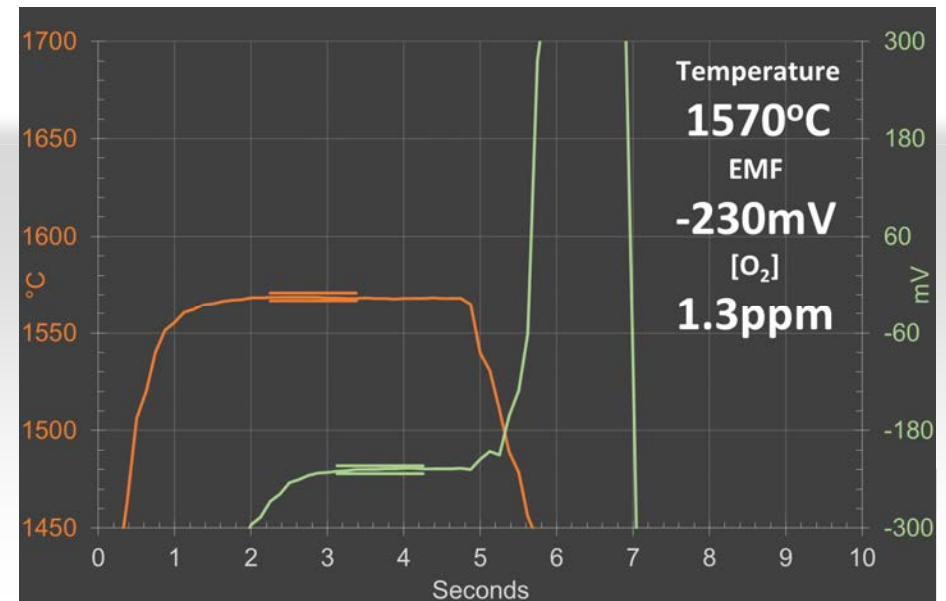
Oxygen measurement - application

Individual examples of ox-probe usage at the steel shop application



High Level of oxygen content during melting in EAF

Low Level of oxygen content after secondary metallurgy treatment



Oxygen measurement

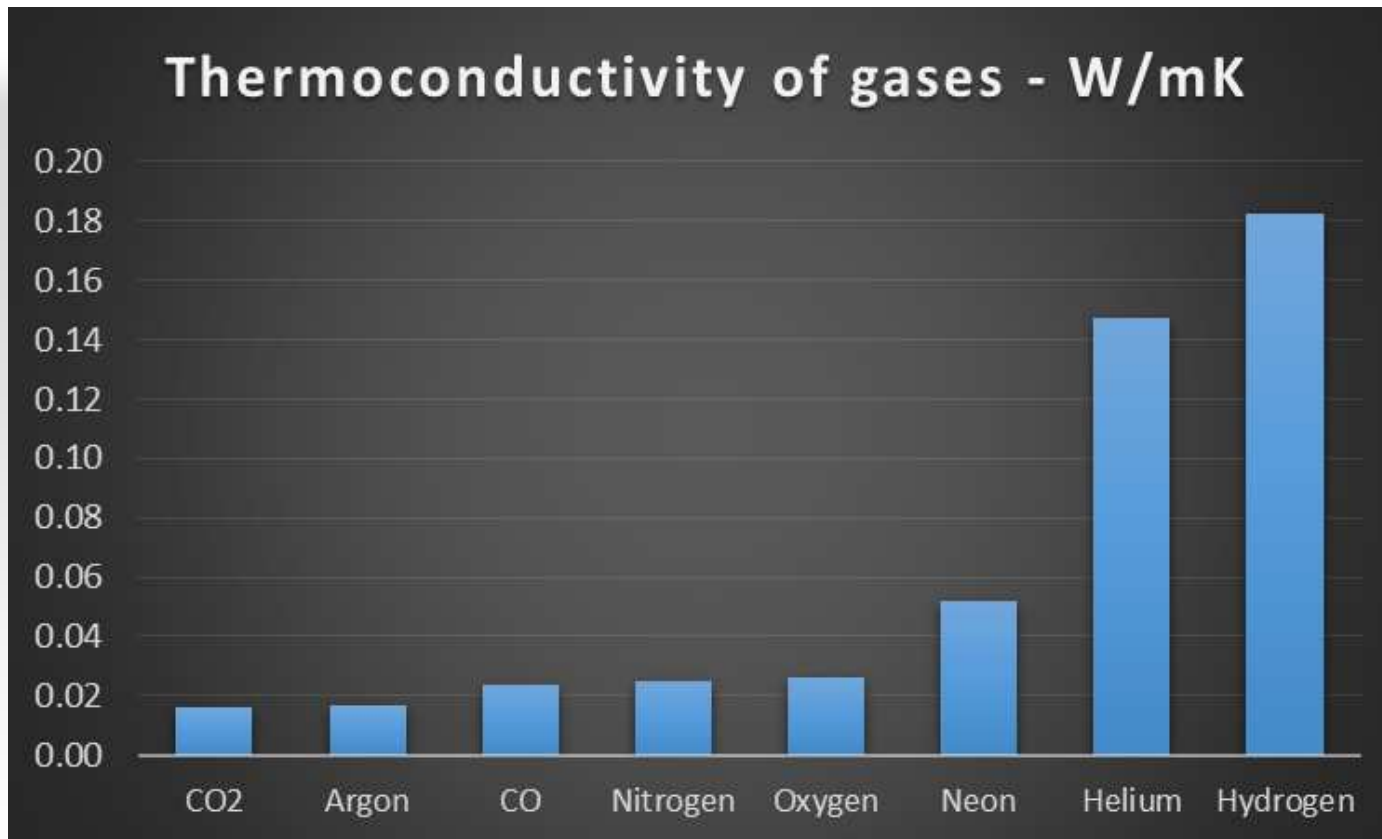
Conclusions

- Electrochemical measurement method (EMF)
- For proper oxygen determination the precise melt temperature measurement is important as well as precise sensor accuracy
- Single dipping probes combine EMF and temperature measurement – Online info about %C and %AL
- High variety of probe types ensure optimum fit to measuring application



Hydrogen measurement

– basic principle THERMOCONDUCTIVITY



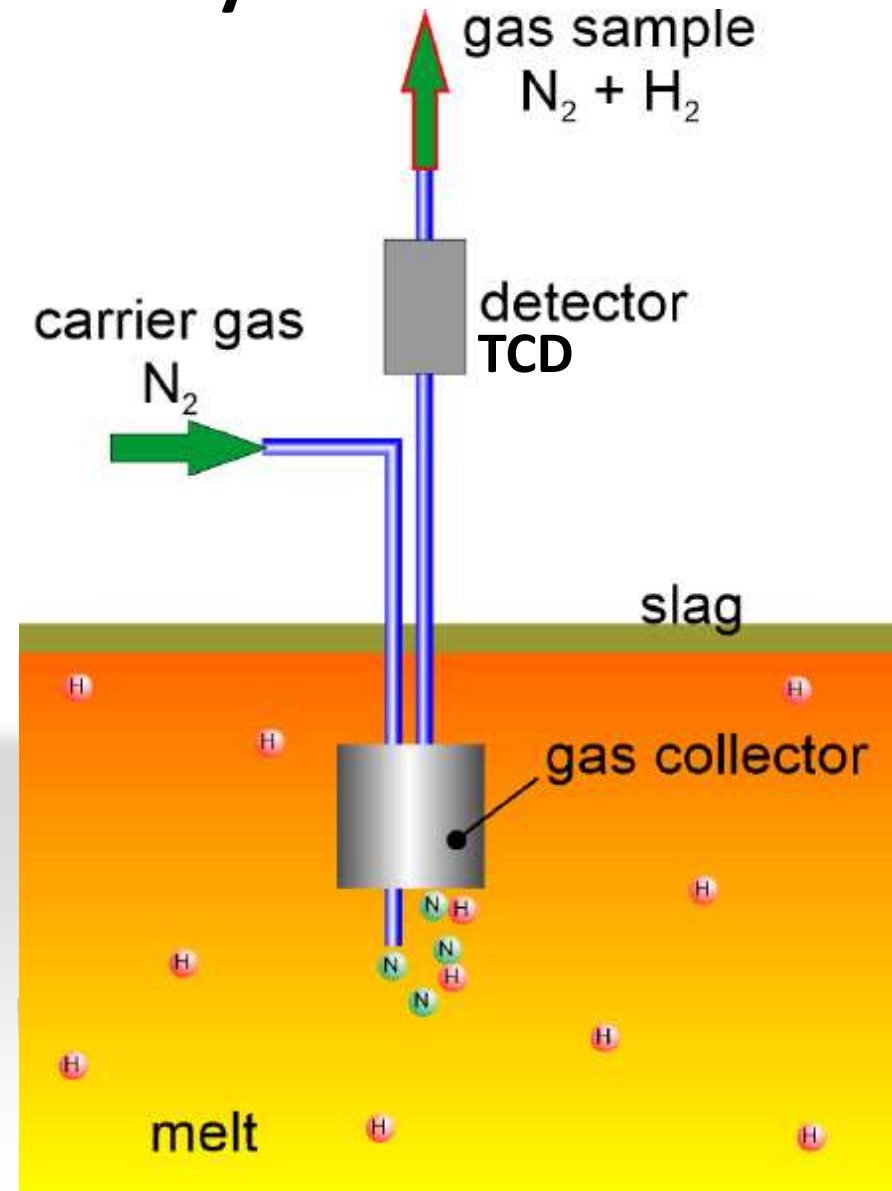
Hydrogen measurement – the theory

Gaseous sampling by using of carrier gas

- Input of carrier gas into the melt
- Collecting of raising carrier gas bubbles
- Determining of the H_2 partial pressure within the collected carrier gas
- Calculation of the H_2 concentration by using the Sievert equation:

$$H = K / f * \sqrt{PH_2}$$

with: H = Hydrogen concentration
 K = equilibrium coefficient
 f = Hydrogen activity
 PH_2 = Hydrogen partial pressure



Hydrogen measurement – the theory

K/f – factor of: $H = K / f * \sqrt{PH_2}$

- K : temperature dependent equilibrium coefficient

$$\text{Log } K_H(T) = - 1900/T + 0.9201$$

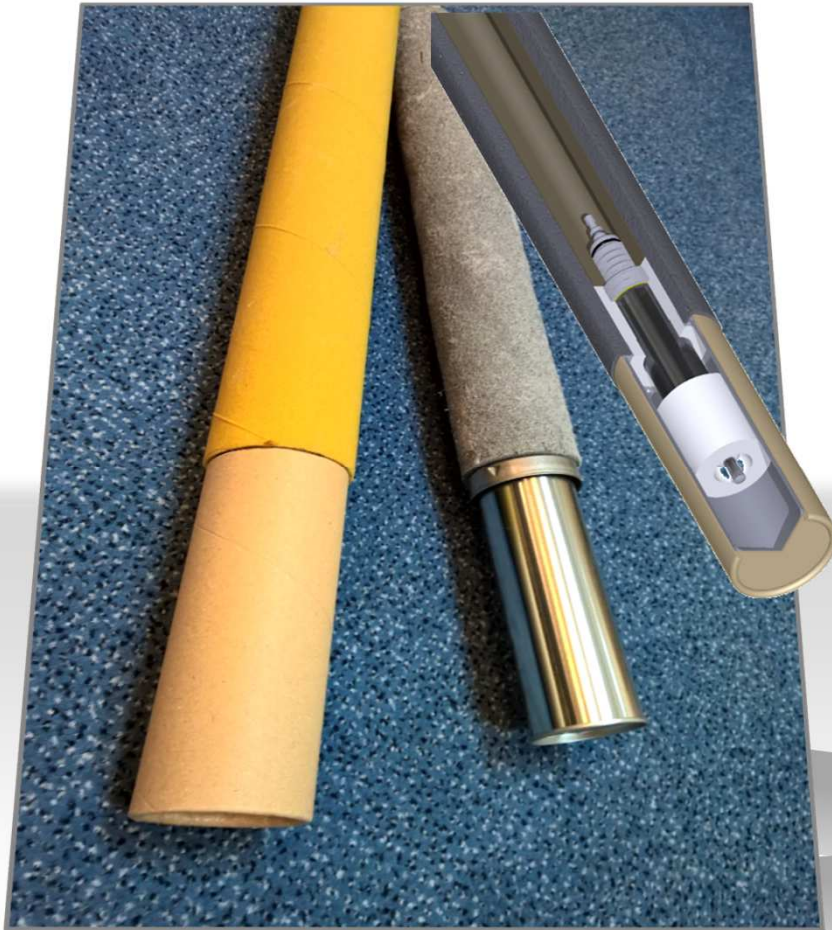
- f : influence of alloying elements on Hydrogen solubility

$$\begin{aligned} \text{Log } f_H = & 0.06 [\%C] - 0.0022 [\%Cr] - 0.0014 [\%Mn] - \\ & 0.0023 [\%Nb] - 0.0016 [\%Ni] + 0.023 [\%Si] \\ & - 0.019 [\%Ti] + 0.0022 [\%Mo] - \\ & 0.0074 [\%V] + 0.0018 [\%Co] \end{aligned}$$

- standard value K/f = 0.750

Hydrogen - Two types of implementation:

Open or closed loop of carrier gas:



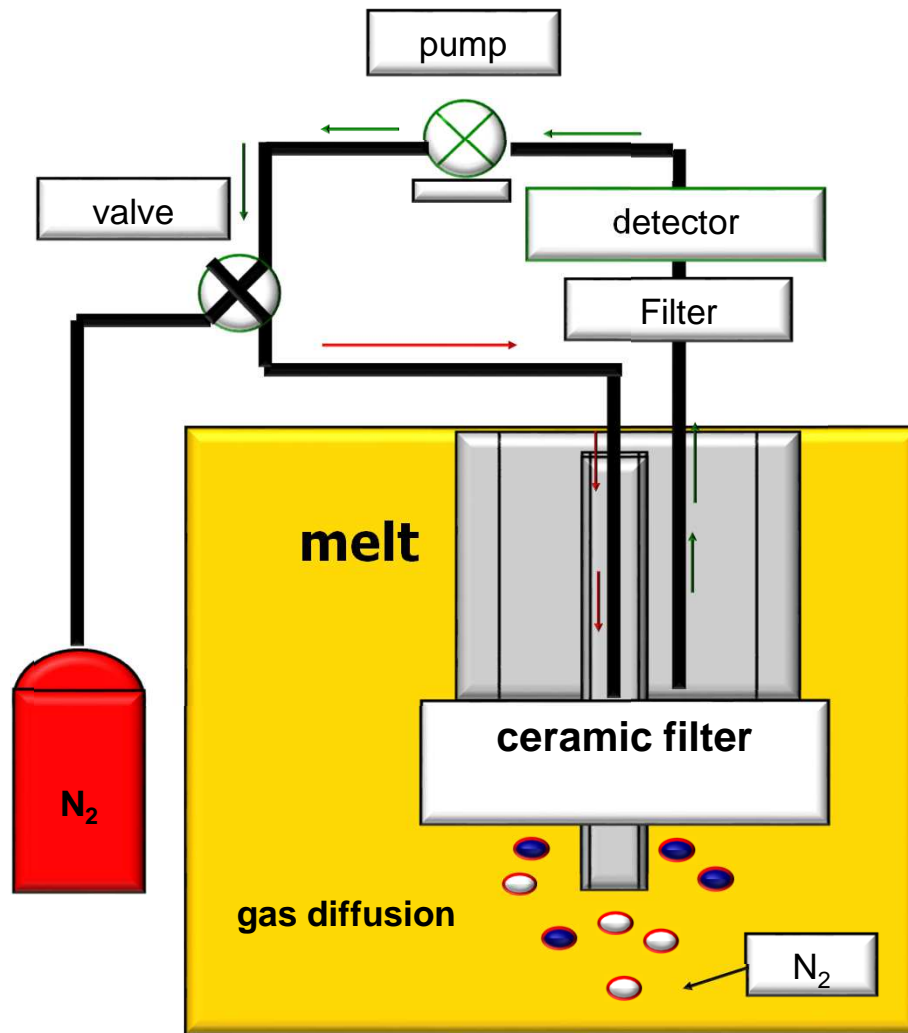
HydroKon CL



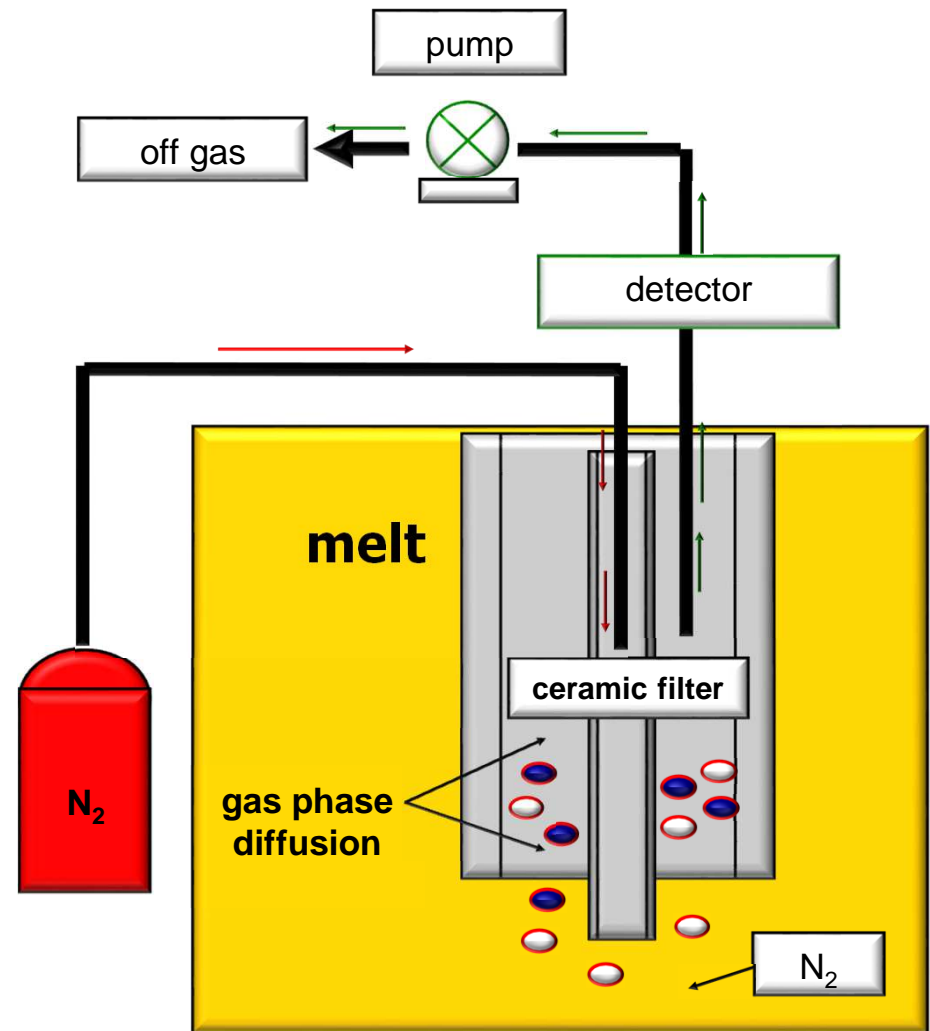
HydroKon OP

Hydrogen - Implementations

Closed loop

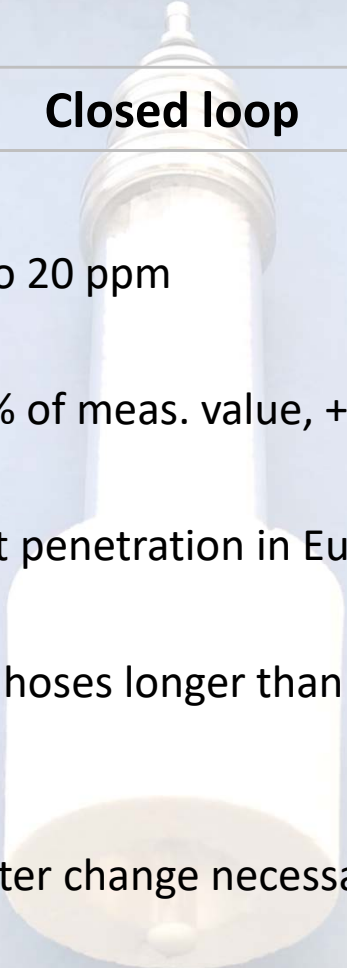
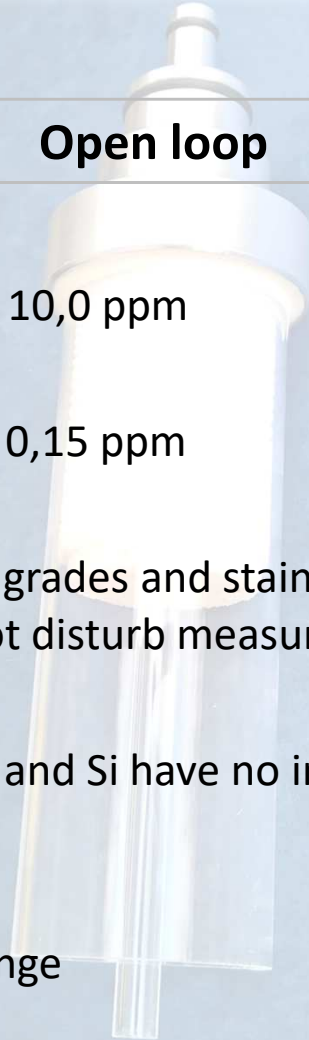


Open loop



Hydrogen measurement – the systems overview

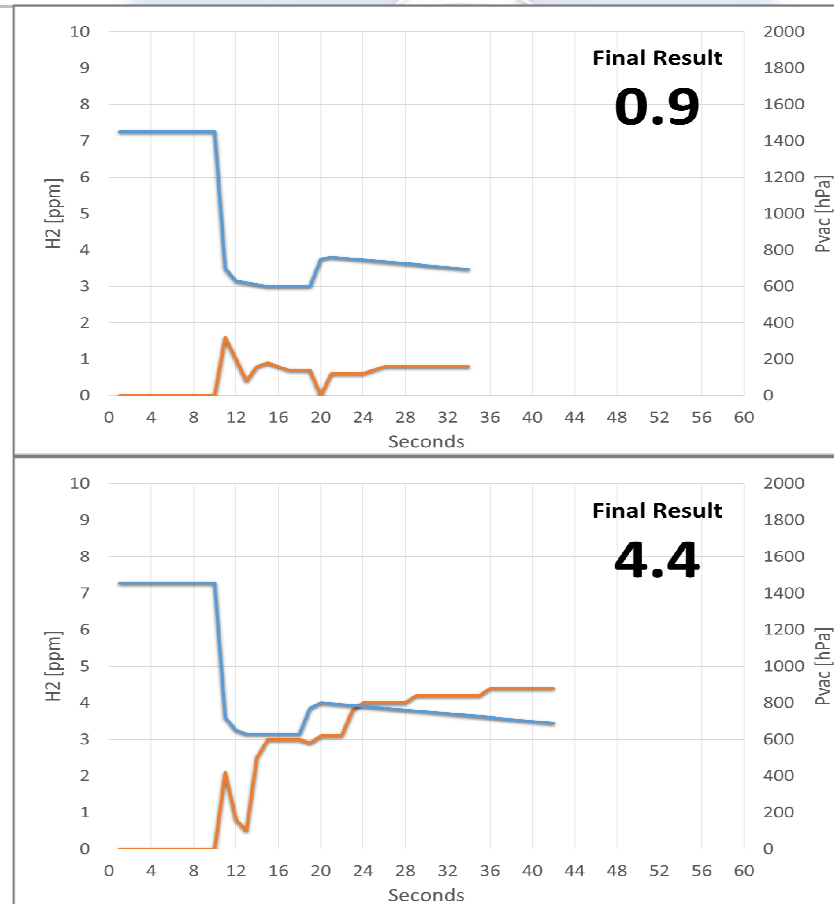
Overview

Closed loop	Open loop
 <ul style="list-style-type: none">▪ Range 0,5 to 20 ppm▪ Accuracy 5% of meas. value, +/- 0,1 ppm▪ High market penetration in Europe▪ Connecting hoses longer than 15 Meter possible▪ Frequent filter change necessary	 <ul style="list-style-type: none">▪ Range 0,1 to 10,0 ppm▪ Accuracy +/- 0,15 ppm▪ High alloyed grades and stainless steel grades do not disturb measuring▪ Ti, S, Pb, Mn and Si have no influence on measuring▪ No filter change

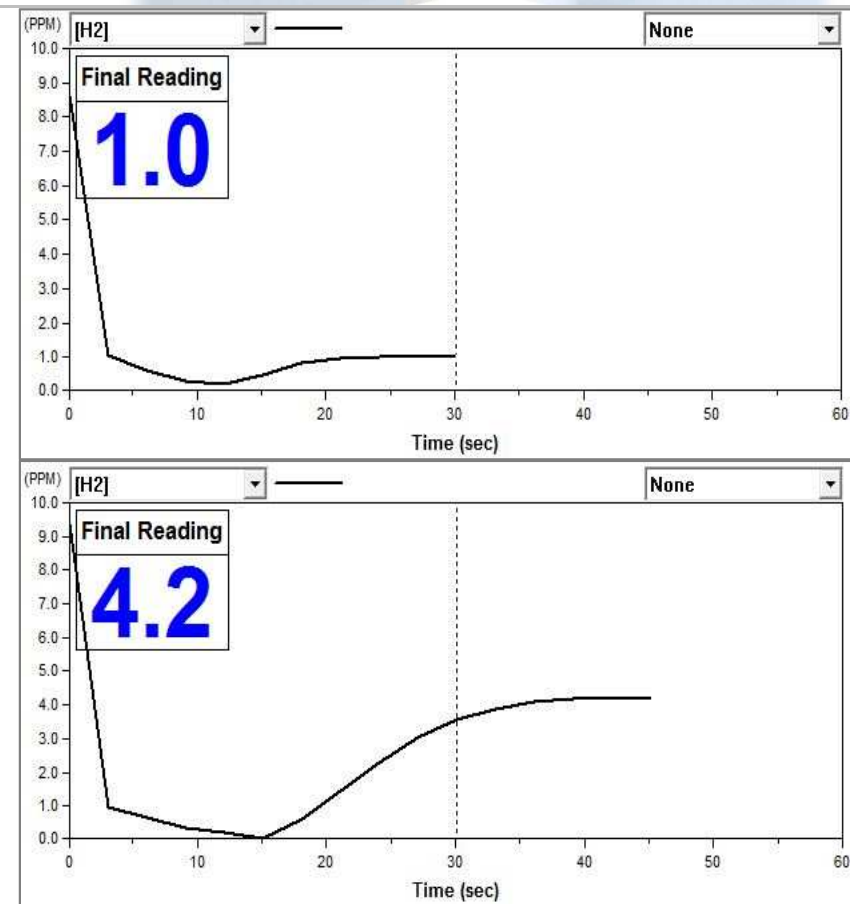
Hydrogen measurement – individual examples

Overview

Closed loop



Open loop



Online measuring of dissolved gasses in high temperature melts

Conclusions

- Dipping probes provide an accurate and reliable online measuring method to determine Oxygen and Hydrogen concentration in the melt
- Allows to save the process operation time for the laboratory check
- The variety of probe types offer an optimal solution for nearly every application
- The deoxidisation process and vacuum degassing can be controlled more precisely with usage of online sensors



Thank you for your attention!

Contact

Minkon GmbH

Heinrich-Hertz-Str. 30-32

D-40699 Erkrath

Telefon: +49 (0)211 209908-0

Telefax: +49 (0)211 209908-90

www.minkon-international.de

